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Spatial and Temporal Variability of Soil Redox Potential, pH and Electrical Conductivity across a Toposequence in the Savanna of West Africa

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Abstract: Soil redox potential is an important factor affecting soil functioning. Yet, very few agronomy studies included soil redox potential in relation to soil processes. The objective of this study was to evaluate the spatial and temporal variation in soil redox potential and to determine the soil parameters affecting its variation. Soil redox potential, soil moisture, soil temperature, pH and bulk electrical conductivity were measured in upland rice fields during two growing seasons at six positions along an upland–lowland continuum, including two positions at the upland, two at the fringe and two at the lowlands in central Côte d'Ivoire (West Africa). The measurements were made at the following soil depths: 3, 8, 20 and 35 cm. Soil redox potential varied between 500 and 700 mV at the upland positions, 400 and 700 mV at the fringe positions and 100 and 750 mV at the lowland positions, and increased with soil depth. Variations in soil redox potential were driven by soil moisture, bulk electrical conductivity and soil organic carbon. We concluded that for proper interpretation of soil redox potential, sampling protocols should systematically include soil pH, moisture and bulk electrical conductivity measurements.

Keywords: ecologies; electrochemical parameters; pH; redox potential; relationships; soil moisture; variation

1. Introduction

Declining soil fertility is a major impediment to agricultural productivity and the main cause of low food production in sub-Saharan Africa [1]. This is the result of poorly adapted production systems such as slash-and-burn farming and too short fallow periods, leading to a continuous degradation of the productive capacity of the land [2–4].

In a context of innovation for sustainable agriculture, there is an urgent need to implement agricultural practices that take into account the processes at stake in soil functioning and are able to



maintain and reinforce soil fertility [5]. This suggests working with effective indicators to understand soil processes to design appropriate cropping systems. Indeed, the dynamics of key indicators and their critical limits are needed to track changes and determine the trends in soil quality improvement or deterioration for various agro-ecological zones [6,7].

Thus, various relevant soil parameters are regularly monitored: pH, a major parameter in many physical–chemical reactions, influences the solubility and bioavailability of many soil minerals [8–10]. Electrical conductivity (EC) is also widely used to characterize various soil properties, such as cation exchange capacity, salinity, nutrient concentration, residual moisture and soil texture [7,11,12]. We can now associate the redox potential (Eh) to the level of oxidation/reduction in a given environment, with soil redox potential values lower than 400 mV indicating a reduced environment. This important parameter, which has long been neglected by agronomists because of the great variability in the measurements [5,13,14], is now accessible after the robustness of the proposed measurement methods [15].

Oxidation–reduction reactions have a profound effect on soil evolution. The chemical evolution of soils is mainly determined by proton and electron fluxes that define the predominant soil mineral fields [16,17] and their solubility [18]. In addition, Eh influences greenhouse gas emissions. Indeed, the processes of decomposition, nitrification, denitrification or methanogenesis, respectively producing carbon dioxide (CO₂), nitrous oxide (N₂O) or methane (CH₄), are typical reductive oxidative reactions that occur through electron exchange between the substrates [19–21]. Furthermore, soil Eh (and pH) largely impacts the vegetation type, and inversely, plants alter soil Eh (and pH), especially in the rhizosphere [5], making the plant essential in the course of the physicochemical processes in the soil.

Thus, with pH, Eh can therefore provide essential information for characterizing soil conditions and its environment. Eh measurements could be used as a key indicator of field processes [5].

In soil the different parameters interact, therefore, soil processes can be better understood when a combination of these relevant parameters is used. However, to do so, there is a need to better understand the spatial and temporal variability of these indicators at the field scale, in order to develop sampling protocols enabling a good control of this variability. Such an approach is an innovation in the West African region.

The objective of this study was therefore to assess the spatial and temporal variation in soil redox potential, pH and EC, and the soil parameters driving their variability in a derived savanna agro-ecological zone of West Africa.

2. Materials and Methods

2.1. Study Area

The study was conducted in rainfed rice fields during two consecutive growing seasons, from June to November 2016 and from June to November 2017, on two close experimental sites at the Africa Rice Center research station M'bé located in the center of Ivory Coast. The station is located 35 km north of Bouaké, between 7.5° and 8.5° North latitude and 4.5° and 5.5° West longitude (Figure 1).

The station is located in the agro-ecological zone of a derived savanna, mainly characterized by a bi-modal rainfall pattern with two rainy seasons and two dry seasons. The first rainy season lasts from March to June and the second rainy season from September to October [22]. Based on the climate data for the period 1994–2014, the average annual rainfall is 1150 mm. Two peaks of rainfall are observed in the study area: the first in June, with a total monthly rainfall amount of 125 mm, and the second in September or October, with a total rainfall amount of 140 mm. The average daily temperatures vary between 30 °C in March and 21 °C in January.

The measurements were made at six row positions along an upland–lowland continuum, including two positions at the upland (Site 1), two at the fringe and two at the lowlands (Site 2). The upland positions (Upland 1 and Upland 2) and the lowland positions (valley bottom) had a relatively gentle

slope (<1% and 2%, respectively) while the fringe and glacis positions had a relatively higher slope, 5% on average (Figure 2).



Figure 1. Location of the experimental site.



Figure 2. Sampling points and measurements.

According to the FAO soil taxonomy, soil at the upland was a Rhodi-humic Arisol; soil along the slope was an Arisol chromi-haplic and soil at the lowland was an Orthi-gleyic Arenosol [22]. Soils of the two sites were characterized before the beginning of the experiment through soil profiles (Figure 3).



Figure 3. Profiles of the soils at the experimental sites.

At each experimental site, soil samples were collected at three soil depths: 0–5 cm, 15–25 cm and 35–45 cm. Soil samples (approximately 250 g) were air-dried at 30–35 °C to reach a water activity close to 0.5, measured with a HP23-AW-SET-14 Rotronic water activity device (PST Group, Bassersdorf, Switzerland) as described in [15]. The physicochemical analyzes were carried out on the fine soil sieved to 2 mm. Organic carbon was measured by the Walkley and Black method by oxidation of the organic material with a mixture of sulfuric acid (H₂SO₄) and potassium dichromate (K₂Cr₂O₇) and titration of the remaining amount of K₂Cr₂O₇ with ferrous sulphate. Granulometric analyzes were performed by sieving and pipetting according to the Robinson pipette method, oxidation of the organic material with hydrogen peroxide (H₂O₂), then dispersion with a solution of sodium hexametaphosphate (NaPO₃)₆ and finally pipetting for the particles [23].

The pH measurements in water (pH- H_2O) were made by extraction with distilled water taking a solid/liquid mass ratio of 1 to 2.5 followed by a reading with a 3520 pH meter (Jenway, Stone, UK). Similarly, the electrical conductivity measurement was carried out in distilled water with a ratio of 1 to 2.5 and reading the conductivity.

Analysis of soil particle size distribution (Table 1) revealed a unique texture on Upland 1 (Site 1): sandy clay loam at horizons 0–5 cm, 15–25 cm and 35–45 cm. On Upland 2, the texture varied from sandy loam in the 0–5 cm horizon to sandy clay loam in the horizons 15–25 cm and 35–45 cm. At Site 2, two textural situations were present on Fringe 1: loamy sand in the horizons 0–5 cm and 15–25 cm and sandy clay loam in horizon 35–45 cm. A unique texture was observed on Fringe 2: loamy sand in horizons 0–5 cm, 15–25 cm and 35–45 cm. On glacis, two textural situations were observed: sandy clay loam in horizons 0–5 cm and 15–25 cm and loamy sand in the horizon 35–45 cm. In the valley bottom, two textural situations also were observed: sandy clay loam in horizons 0–5 cm and 15–25 cm and clay loam in horizon 35–45 cm.

	Soil Depth	рН (H ₂ 0)	Organic Carbon (%)	Electrical Conductivity (µS/cm)	Clay (%)	Silt (%)	Sand (%)	Texture	
Upland 1	0–5 cm	6.48	3.45	119.0	23.25	17.75	59.0	Sandy clay loam	
	15–25 cm	6.70	2.41	54.1	28.25	12.25	59.5	Sandy clay loam	
	35–45 cm	6.70	1.37	47.3	47.3 31.46 10.50 58.0 Sand		Sandy clay loam		
Upland 2	0–5 cm	6.27	2.62	86.4	16.21	11.25	72.5	Sandy loam	
	15–25 cm	6.40	1.43	32.2	26.00	8.74	65.3	Sandy clay loam	
	35–45 cm	6.31	1.03	28.6	30.77	8.22	61.0	Sandy clay loam	
Fringe 1	0–5 cm	6.59	2.15	89.4	12.00	7.50	80.5	Loamy sand	
	15–25 cm	6.59	1.43	61.0	17.50	7.00	75.5	Loamy sand	
	35–45 cm	6.61	1.05	55.3	24.00	6.50	69.5	Sandy clay loam	
Fringe 2	0–5 cm	6.57	1.75	84.4	5.93	7.50	86.5	Loamy sand	
	15–25 cm	6.76	0.92	35.9	6.50	6.85	86.7	Loamy sand	
	35–45 cm	6.74	0.56	43.2	6.32	7.00	86.7	Loamy sand	
Glacis	0–5 cm	6.24	2.92	143.3	22.25	16.25	61.5	Sandy clay loam	
	15–25 cm	6.56	1.20	54.5	23.50	11.50	65.0	Sandy clay loam	
	35–45 cm	6.47	0.32	34.0	14.00	6.50	79.5	Loamy sand	
Valley bottom	0–5 cm	6.32	3.56	189.2	22.00	24.25	53.8	Sandy clay loam	
	15–25 cm	6.78	1.20	92.8	28.67	18.33	53.0	Sandy clay loam	
	35–45 cm	6.93	0.63	114.5	37.00	18.75	5 44.3 Clay loam		

Table 1. Mean values (n = 4) of the soil characteristics at different depths at the experimental sites. Detailed values showing their spatial variability are given in Figure S1 in the Supplementary Material.

All the experimental sites were slightly acidic (6.24–6.93). Soil organic carbon and electrical conductivity decreased with soil depth (Table 1).

2.2. Experimental Design

The experiments were conducted at 2 sites located along the upland–lowland continuum. Both sites were reclaimed in 2015, after over 12 years of being fallow but with frequent burning by wildfires. On Site 1, the upper part of the toposequence located on the plateau, the vegetation before reclamation consisted of tall (up to 4 m high) perennial grasses dominated by *Pennisetum purpureum Schumach., Andropogon gayanus Kunth* and *Hyparrhenia rufa* (*Nees*) *Stapf,* mixed with other species such as *Digitaria horizontalis Willd., Mucuna pruriens* (L.) *DC., Aeschynomene histrix Poir., 1816, Tridax procumbens L.* or *Spermacoce stachydea DC.* The vegetation on Site 2 showed a strong gradient across the toposequence: the vegetation of the sandy-loamy fringe was largely dominated by *Imperata cylindrica* (L.) *Raeusch,* with ocally a few species such as *Hyparrhenia involucrata Stapf* or *Digitaria horizontalis Willd.* In between, on the "glacis", the vegetation consisted of a diverse mixture of *Imperata cylindrica* (L.) *Raeusch., Leersia hexandra Sw., Cyperus* spp. (especially *C. ferrax* and *C. esculentus*), *Centrosema* spp., *Phyllanthus amarus Schumach. & Thonn., Aspilia bussei bussei O. Hoffm. & Muschl, Eclipta prostrata* (L.) L., etc. In the "bas-fond", the lowest part of the toposequence, the vegetation was largely dominated by *Leersia hexandra Sw.*

Land reclamation started in 2015, with the destruction of perennial grasses through crushing and application of herbicide (glyphosate 1800 g ha⁻¹) once re-growing plants were 30 cm tall. Rice was then cultivated for the first time after ploughing and harrowing, and sawn in June 2015. During the two consecutive growing seasons (June–November 2016 and June–November 2017), land at each experimental site was firstly plowed to a depth of 30 cm in March and secondly plowed to a depth of 30 cm in May to remove all weeds and properly harrow the field. A rice variety able to grow in a wide range of hydrological conditions was selected as the crop in this study. The upland rice variety NERICA 4, with a broad adaptation range [24], was grown; it was planted in pods at a 0.3 m × 0.3 m spacing—5 to 8 grains per pod—with a basal fertilizer application of 200 kg ha-1 of NPK (12–24–18), equivalent to

24 kg N ha⁻¹, 48 kg P ha⁻¹ and 36 kg K ha⁻¹, a first top-dressing fertilizer (urea at 87 kg ha⁻¹ at 30 days after sowing (DAS)) and a second-top dressing fertilizer (urea at 87 kg ha⁻¹ at 60 DAS). Seeds were treated against fungi and insects and herbicide (oxadiazon, 1000 g ha⁻¹) at sowing. No pesticide was applied afterwards. The experimental design was a randomized complete block with 4 replicates.

2.3. Measurement of Electrochemical Soil Parameters

2.3.1. Measurement of Soil Redox Potential (Eh)

Soil redox potential was measured following [15], with a WTW3110 voltmeter (WTW, Weilheim, Germany) (with an input resistance greater than 5×10^{12} Ohm), an radiometer analytical Ag/AgCl (KCL 3M) reference electrodes Ref 321 and a 40-cm-long, handmade probe produced by Paleoterra (Netherlands), made of tubes of 8 mm in diameter epoxy reinforced with fiberglass in which was integrated platinum ((Pt) pure redox sensors (99.95%). The probe was connected to a cable filled and sealed with epoxy to form a sturdy and waterproof assembly that can be hammered into the ground. The Pt sensors were distributed on the probe to obtain measurements at depths of 3 cm, 8 cm, 20 cm and 35 cm. Each Pt sensor was connected to its own wire inside the cable. Soil redox potential (Eh) was measured simultaneously on the different depths by connecting each Pt sensor to a different voltmeter and reference electrode.

Before each series of measurements, the electrodes were cleaned with very fine sand paper and calibrated regularly following [15]. The electrodes rinsed with distilled water were then calibrated in a 220 mV Zobell buffer solution before being rinsed and recalibrated in a 220 mV Zobell buffer diluted to 1% in 0.1M KCl. Finally, the electrodes were rinsed and wiped with paper and used for a series of 10 to 15 measurements.

The redox potential was measured by vertically planting the redox probe at the desired depth into the ground, using a hammer. The Ag/AgCl reference electrodes associated with the different Pt sensors were gently inserted into the soil at 2–3 cm near the redox probe (radius 5–15 cm). The E value for each depth was the E (in mV), indicated on the voltmeter that was associated after 1 min with no change in the E unit. After being measured, according to the Ag/AgCl reference electrode, all potentials vs. AG/AgCl (E) were transformed to give Eh according to the normal hydrogen electrode (NHE) with the relationship between the Ag/AgCl reference, temperature and NHE provided by the radiometer instrument.

The first measurement at the beginning of a series, after the electrode testing and cleaning, have always been considered as a "test" and were not included in the data analysis [15].

On the plots, measurements were collected as far as possible at a frequency of 15 days at each site, between 08:00 and 11:00 a.m. to account for the variability of redox due to changes in soil temperatures during the day. In case of rain, the measurements were suspended for 24 to 48 h.

In order to better define the redox conditions of the soils studied, Eh can be adjusted to pH 7 [15,25]

$$Eh_{ph7} = Eh - \frac{RT}{F} \times Ln10 \times (7 - pH)$$
⁽¹⁾

where Eh_{pH7} is the soil redox potential at pH = 7, *R* is the perfect gas constant (8.314 JK⁻¹ mol⁻¹), *T* is the temperature in Kelvin (25 °C = 298, K) and *F* the Faraday constant (96,485 C mol⁻¹ = 1 F). At 25 °C, Eh_{pH7} = Eh-0.059 × (pH 7).

In this study, we used the notion of pe + pH, which is equivalent to Eh_{pH7} , as shown by [26] as

$$Eh_{pH7} = \frac{RT}{F} \times Ln10 \times (pe + pH - 7)$$
⁽²⁾

or

$$pe + pH = \frac{Eh_{pH7} x F}{RT x Ln10} + 7$$
 (3)

2.3.2. Measurement of Soil pH and Temperature

The soil pH and temperature (°C) were measured at depths of 3 cm, 8 cm, 20 cm and 35 cm with a Field Scout Soil Stick pH Meter (Spectrum Technologies Inc., Aurora, IL, USA), which is equipped with a flat surface electrode and simultaneously displays pH and temperature. Before each series of measurements, the pH meter soil stick was calibrated with pH 4 and pH 7 solutions. The pH meter was regularly cleaned with water between measurements.

2.3.3. Measurement of Soil Humidity and Electrical Conductivity

Volumetric soil moisture and bulk electrical conductivity (Ecb) were measured at depths of 3 cm, 8 cm, 20 cm and 35 cm with a portable HH2 moisture meter (Delta-T Ltd. device, Cambrigde, UK) and a portable WET-2 sensor (Delta-T Ltd. device, Cambrigde, UK).

2.4. Statistical Analyses

Statistical tests, models and figures were done with Statistica 7.1 and R software. An analysis of variance was conducted on the data collected. Mean values were tested for significant differences using a least significant difference (LSD) or standard error of difference (SED) test. The probability level of 0.05 was designated as significant. Correlation circles and multi-linear regression analyses were conducted to identify the associations and relationships between the soil redox potential and the soil properties at each experimental site and soil depth.

3. Results

3.1. Rainfall and Air Temperature

The cumulative rainfall amount was higher during the growing season (June–December) of 2017 (785 mm) than during the growing period (June–December) of 2016 (540 mm) (Figure 3). The rainfall amount was low at the beginning of the growing season in June (45 mm) and reached its peak in September with 178 mm and 215 mm in 2016 and 2017, respectively. The rainfall was more evenly distributed in 2017 than in 2016, with 57 rainfall events in 2017 compared with 51 rainfall events in 2016. During the growing season, the average daily air temperature slightly varied between 21 and 26 °C in both years. At the experimental sites, there was a seasonal variation in the daily temperature with the lowest temperatures (21–26 °C) recorded during the period June–November and the highest temperatures (30 °C) recorded during the period March–May. The average daily temperature was about 25 °C in 2016 and 2017 (Figure 4).



Figure 4. Daily rainfall (mm) and daily air average temperature (°C) during the 2016 and 2017 crop seasons at the experimental station.

3.2. Soil Moisture

The evolution of the soil moisture in the uplands (Upland 1 and Upland 2), fringes (Fringe 1 and Fringe 2) and lowlands (glacis and valley bottom) during the 2016 and 2017 growing seasons are shown in Figure 4. The average soil moisture was significantly higher in 2017 ($0.21 \text{ m}^3 \text{ m}^{-3}$) than in 2016 ($0.17 \text{ m}^3 \text{ m}^{-3}$) (Table 2). The mean soil moisture during the two growing seasons was significantly higher in the lowland ecology ($0.28 \text{ m}^3 \text{ m}^{-3}$ and $0.22 \text{ m}^3 \text{ m}^{-3}$, respectively, for valley bottom and glacis) than on the uplands ($0.15 \text{ m}^3 \text{ m}^{-3}$ and $0.12 \text{ m}^3 \text{ m}^{-3}$, respectively, for Upland 1 and Upland 2) and fringes ($0.11 \text{ m}^3 \text{ m}^{-3}$ and $0.08 \text{ m}^3 \text{ m}^{-3}$, respectively, for Fringe 1 and Fringe 2). Soil moisture was higher in the deepest soil horizons than at the surface horizon (Table 2). Soil moisture also significantly varied within the growing season and fluctuated with the rainfall amount. In general, the soil moisture was higher in the months of September and October when the highest rainfall amount was recorded (Figure 5).

Treatments	Temperature (°C)	pН	Moisture (m ³ .m ⁻³)	ECb (mS.cm ⁻¹)	Eh (mV)	Pe + pH
Year (Y)						
2016	28.2 ^a	5.94 ^a	0.17 ^a	0.14 ^a	560 ^a	15.35 ^a
2017	29.1 ^b	5.99 ^a	0.21 ^b	0.18 ^b	562 ^a	15.34 ^a
SED (main year effect)	0.1	ns	0.01	0.01	ns	ns
Site (S)						
Upland 1	29.5 ^c	6.40 ^d	0.15 ^d	0.09 ^b	570 ^{b,c}	15.90 ^d
Upland 2	28.2 ^a	5.90 b	0.12 ^c	0.06 ^{a,b}	572 ^{b,c}	15.42 ^{b,c}
Fringe 1	29.0 ^{b,c}	6.06 ^c	0.11 ^b	0.04 ^{a,b}	556 ^{a,b}	15.32 ^b
Fringe 2	29.1 ^c	6.13 ^c	0.08 ^a	0.03 ^a	558 ^{a,b,c}	15.43 ^{b,c}
Glacis	28.1 ^a	5.70 ^a	0.22 ^e	0.12 ^c	544 ^a	14.66 ^a
Valley bottom	28.7 ^b	6.12 ^c	0.28 ^f	0.38 ^d	574 ^c	15.71 ^{c,d}
SED (main site effects)	0.4	0.1	0.02	0.04	16	0.27
Depth (D)						
3 cm	29.0 ^c	5.94 ^a	0.17 ^a	0.12 ^a	550 ^a	15.12 ^a
8 cm	28.5 ^b	5.98 ^a	0.19 ^b	0.14 ^a	564 ^b	15.42 ^b
20 cm	28.4 ^a	5.98 ^a	0.20 ^c	0.18 ^b	566 ^b	15.44 ^b
35 cm	29.0 ^c	5.99 ^a	0.23 ^d	0.29 ^c	569 ^b	15.47 ^b
SED (main depths effects)	0.2	ns	0.01	0.03	10	0.2
Source of Variance			<i>p-</i> Valu	e		
Year (Y)	< 0.001	0.06	< 0.001	< 0.001	0.40	0.63
Site (S)	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001
Depth (D)	< 0.001	0.45	< 0.001	< 0.001	0.03	0.01
Y×S	< 0.001	0.04	0.69	0.85	0.41	0.12
Y×D	0.49	0.86	0.37	0.02	< 0.001	< 0.001
S×D	0.99	0.77	0.95	0.99	0.91	0.94
$Y \times S \times D$	0.99	0.05	0.05	< 0.001	0.03	0.03

Table 2. Average soil electrochemical characteristics at different soil depths of the experimental sites.

ECb: bulk electrical conductivity; Eh: soil redox potential; pH: soil pH; Pe + pH: soil pe + pH. SED: standard error of a difference. ^{a,b,c,d,e,f}—letters in a column within a set indicate significant difference at p < 0.05 by the least significant difference test. ns: not significant.



Figure 5. Soil moisture at different soil depths during the 2016 and 2017 growing seasons. Error bars indicate the standard errors of the mean.

3.3. Soil Temperature Variation

Soil temperature varied greatly, depending on the year of the experiment, the location of the site and the depth of the soil (Figure S2). On average, soil temperature was higher in 2017 (29.1 °C) than in 2016 (28.2 °C). The greatest amplitudes of temperature variation were recorded at Upland 2, between 23.5 °C and 37 °C in 2016, and at Fringe 1, between 25.8 °C and 35.1 °C in 2017. The smallest amplitudes were registered at the glacis, between 25.5 °C and 31 °C in 2016, and at the valley bottom, between 25.1 °C and 32.8 °C in 2017. The lowest average temperature was recorded at the glacis, 28.1 °C, and the mean highest temperature was recorded in Upland 1 (29.5 °C).

There was also a significant year-by-site interaction effect on soil temperature. In 2016, the lowest soil temperature was recorded at Upland 2 (23.5 °C), while in 2017, the lowest soil temperature was found at the glacis (25.1 °C). On average, the soil temperature was higher at a 3 cm depth and decreased with soil depth (Table 2).

3.4. Electrical Conductivity Variation

The evolution of the bulk electrical conductivity (Ecb) of soils is presented in Figure 6. The bulk electrical conductivity varied significantly with the location of the site. Conductivity was significantly higher in 2017 (Table 2). Lowland ecology recorded the highest averages (0.38 mS.cm⁻¹ and 0.12 mS.cm⁻¹, respectively, for valley bottom and glacis). The fringes recorded the lowest averages (0.04 mS.cm⁻¹ and 0.03 mS.cm⁻¹, respectively, for Fringe 1 and Fringe 2). On the upland, the averages of 0.09 mS.cm⁻¹ and 0.06 mS.cm⁻¹ were recorded for Upland 1 and Upland 2, respectively (Table 2). In the valley bottom where the highest Ecb was observed, there was a strong fluctuation, ranging from 0.14 mS.cm⁻¹ to 0.48 mS.cm⁻¹ in 2016 and from 0.14 to 0.77 mS.cm⁻¹ in 2017. The lowest soil conductivity was at Fringe 2 and ranged from 0.01 mS.cm⁻¹ to 0.1 mS.cm⁻¹ in 2016 and 0.00 mS.cm⁻¹ to 0.05 mS.cm⁻¹ in 2017. At all sites, the lowest conductivity was recorded at a depth of 3 cm with an average of 0.12 mS.cm⁻¹ and 0.29 mS.cm⁻¹ (Figure 6).

	2016	2017			
Upland 1	Ecb_3 Ecb_8 Ecb_20	Ecb_3 Ecb_3 Ecb_8 Ecb_35 E			
Upland 2	Ecb_3 Ecb_8 Ecb_20	Ecb_3 Ecb_8 1.40 Ecb_20 Ecb_35 Ec			

Figure 6. Cont.



Figure 6. Soil bulk conductivity at different soil depths during the 2016 and 2017 growing seasons. Error bars indicate the standard errors of the mean.

3.5. Soil pH Variation

There was no significant year-by-site location interaction effect on soil pH (Table 2 and Figure 7). The lowest pH was observed at the glacis (5.70), followed by Upland 2 (5.90). The highest pH was observed at Upland 1 (6.40). The soils of Fringe 1, Fringe 2 and the valley bottom had a pH 6.06, 6.13 and 6.12, respectively.

The smallest fluctuation amplitudes during the crop cycle were observed at Upland 1 (0.26 and 0.62, respectively, in 2016 and 2017) and then followed the fluctuation amplitudes observed in 2016 and 2017 on Fringe 1 (0.9 and 0.42), Upland 2 (1.07 and 0.96) and Fringe 2 (0.97 and 1.56). The highest fluctuation amplitudes were recorded in the lowland ecology on the glacis (1.2 and 1.91) and valley bottom (1.34 and 1.52) (Figure 7).



Figure 7. Soil pH at different soil depths during the 2016 and 2017 growing seasons. Error bars indicate the standard errors of the mean.

3.6. Soil Redox Potential Variation

Figure 8 presents the ecological dynamics of the redox potential for the uplands (Upland 1 and Upland 2), fringes (Fringe 1 and Fringe 2) and lowlands (glacis and valley bottom) soils during the growing season (June to December) of the 2016 and 2017 seasons. It was observed that, at the beginning of each season, the soil redox potential was around 500 mV before fluctuating during the growing season. This dynamic varied according to the site and the depth of the measurement. However, there was no significant difference between the average soil redox potential over the two years of study (Table 2).



Figure 8. Cont.



Figure 8. Soil redox potential at different soil depths during the 2016 and 2017 growing seasons. Error bars indicate the standard errors of the mean.

The mean Eh was significantly lower at the glacis (544 mV) and the valley bottom had the highest Eh average (574 mV), followed by the uplands (570 and 572 mV for Upland 1 and Upland 2, respectively) and fringe soils (556 and 558 mV for Fringe 1 and Fringe 2, respectively).

During the 2016 growing season, the redox potential of the upland soils was more stable. The variations over the entire cycle remained in the range of 516 to 595 mV (Upland 1) and 520 to 632 mV (Upland 2) for the three depths of 3, 8 and 20 cm. The lowest average redox potential was recorded at a depth 3 cm with mean values of 536 and 535 mV for Upland 1 and Upland 2, respectively, and the redox variations at the fringe soils ranged from 479 to 591 mV (Fringe 1) and from 464 to 591 mV (Fringe 2). A depth of 3 cm recorded the lowest average redox, 504 mV and 518 mV, respectively, for Fringe 1 and Fringe 2. At the lowland ecology, redox fluctuations in the soil were 544 and 628 mV for the valley bottom and more pronounced for the glacis (381 to 614 mV), for depths of 3, 8 and 20 mV in 2016. A depth of 20 cm recorded the mean lowest redox potential (521 mV) at the glacis.

In 2017, the upland soils remained relatively stable throughout the cycle, with the redox potential varying from 540 to 601 mV and from 521 to 652 mV for Upland 1 and Upland 2, respectively, over all the depths—3, 8, 20 and 35 cm. On these soils, the lowest average redox potential was recorded at a depth of 3 cm.

Redox potential in the fringe soils varied from 512 to 648 mV and from 500 to 662 mV for Fringe 1 and Fringe 2, respectively. A depth of 3 cm had the lowest redox average and its relatively small amplitude of variation was 23 mV and 50 mV for Fringe 1 and Fringe 2, respectively.

At the lowland soils, the redox potential was particularly subject to repeated variations influenced by the rainfall regime. The amplitude of the variations was large for the depths 3, 8, 20 and 35 cm, with the redox potential varying from 249 to 714 mV and from 466 to 639 mV for the glacis and the valley bottom, respectively. In mid-October 2017, the redox potential of the 8 and 20 cm depths of the glacis dropped below 400 mV. The lowest redox potential was recorded at a depth of 20 cm: 250 mV.

During the growing season, the amplitudes of the redox potential variations of the studied depths were particularly high on glacis, with 235, 355, 416 and 443 mV for the respective depths of 3, 8, 20 and 35 cm. On the valley bottom soils, the amplitudes were 82, 145, 207 and 185mV, respectively, for the depths 3, 8, 20 and 30 cm.

On the lowland ecology, the lowest mean redox potential levels were inversely recorded at a depth of 35 cm with 528 mV and 557 mV, respectively, for the glacis and valley bottom (Figure 8).

3.7. Variation in Soil pe + pH

The mean soil pe + pH remained stable over the two years of study (Table 2). Soil pe + pH was significantly lower on the glacis (14.66) and higher on Upland 1 (15.90). The pe + pH averages of the other sites were 15.32, 15.42. 15.43 and 15.71 for Fringe 1, Upland 2, Fringe 2 and valley bottom, respectively.

There was a lower fluctuation of soil pe + pH at the upland soils (1.48–1.36 in 2016 and 1.37–2.26 in 2017 for Upland 1 and Upland 2, respectively) followed by fluctuations observed on the fringe (1.78–1.9 in 2016 and 2.1–2.68 in 2017 for Fringe 1 and Fringe 2, respectively). Soils at the lowland ecology recorded the highest fluctuations (3.93–2.45 in 2016 and 7.62–3.71 in 2017, for glacis and the valley bottom, respectively) (Figure 9). The mean pe + pH was significantly lower at a depth of 3 cm (15.12) compared to the mean of the depths 8, 20 and 35 cm with 15.42, 15.44 and 15.47, respectively (Figure 9).



Figure 9. Cont.





Figure 9. Soil pe + pH at different soil depths during the 2016 and 2017 growing seasons. Error bars indicate the standard errors of the mean.

3.8. Relationships between Soil Potential Redox and Other Soil Electrochemical Properties

Relationships between the soil electrochemical and water parameters were studied from the averages of the data collected at the measurement points on the different sites during the two growing seasons (Figure 10).

Figure 10a shows a two-phase effect of moisture on the soil redox potential. A phase of increase of the redox potential is observed, with the increase of the soil moisture at the depths 3 cm, 8 cm and 20 cm reaching a maximum of $0.20 \text{ m}^3 \text{.m}^{-3}$ soil moisture. Beyond this threshold value, a decay phase of Eh is observed over the three depths to reach values of 545–538 mV at 0.30–0.36 m³.m⁻³ of humidity. Quadratic functions to better reflect the effect of moisture, highlight a gradient of Eh between depths of 3 cm, 8 cm and 20 cm. Significant (<0.001) quadratic correlations were also established between Eh and the soil moisture content for each depth, with R² correlation coefficients of 0.71, 0.51 and 0.54, respectively, for depths of 3 cm, 8 cm and 20 cm.

A strong and positive correlation (p < 0.001) exists independently of depths (3 cm, 8 cm and 20 cm) between moisture and bulk electrical conductivity (Ecb); the R² coefficients are between 0.93 and 0.97 (Figure 10b). As a result, an effect similar to that of moisture was observed between Ecb and Eh. A phase of increase of Eh with Ecb was observed, followed by a decay phase after 0.2 mS.cm⁻¹ (Figure 10c). A depth gradient (3 cm, 8 cm and 20 cm) of Eh was observed on each phase. The correlations are significant, and their coefficients are 0.79, 0.46 and 0.36, respectively, for depths 3 cm, 8 cm and 20 cm.

In addition, soil moisture content showed a positive and significant correlation at a depth of 3 cm with soil organic carbon (SOC) content ($R^2 = 0.68$); consequently, at this depth a positive effect of the carbon level on Eh was observed with a significant correlation and a correlation coefficient R^2 of 0.61 (Figure 10d,e). Such correlations are not found at 20 cm depth. To find significant correlations, the soils must be considered separately according to upland, fringe and lowland ecologies (Figure S3).

Globally, correlations between Eh and pH are poor (R^2 of 0.11, 0.11 and 0.22, respectively, for the depths 3 cm, 8 cm and 20 cm, observed in Figure 10f) and not significant. However, considering the upland, fringe and lowland ecologies separately, stronger correlations are obtained (Figure S4).



Figure 10. Relationship between soil moisture and Eh (**a**). Correlation of soil moisture and electrical conductivity and its consequence on Eh. (**b**,**c**) Effect of soil organic carbon (SOC) on soil moisture and its consequence on Eh (**d**,**e**). Relationship between pH and Eh (**f**). The regression equations and correlation coefficients (\mathbb{R}^2 and p values) are determined for each depth.

4. Discussion

Similar redox potentials were observed for the two seasons of the study. The average redox potential was significantly different from one soil type to another and the standard errors at a given day and in a given site were rather low. This tends to confirm the reliability of the method of measurement of soil redox potential.

For all sites, soil Eh ranged from 106 to 717 mV. These values are in the fluctuation range of the soil redox values of -300 to +900 mV given in the literature [5,27]. Soil pH measured directly in field conditions was most of the time lower than the soil pH measured in laboratory conditions in a water extract (1:2.5 v/v). As for Eh, pH showed great variation in time, especially in 2017 in the lower part of the toposequence (glacis and valley bottom), where the soil pH could vary by more than 1.5 units within a few weeks. Similarly, ECb greatly varied, increasing with soil moisture.

Globally, these results confirm the key role of soil moisture in soil functioning, as it is central to the dynamics and relationships between soil electrochemical parameters.

In our study area and conditions, soil Eh variability could be mainly attributed to humidity, with a quadratic relationship between soil moisture and soil Eh. The raise in Eh with the raise in humidity in dry conditions reveals the chemical functioning of a soil with poor biological activity: low redox potential in dry soils are attributed to the loss of colloidal water that retains oxygen, and thus the loss of oxygen in a purely chemical process [15]. In a biologically active soil, the raise in soil moisture would favor the development of microorganisms, which consume oxygen and, as a consequence, would lower the Eh [28,29]. Although microbial activity has not been measured in this study, a low biological activity can be expected in these soils due to the frequent fires they endure and the cropping practices with soil tillage.

In all cases, with increasing water content, when the soil reaches saturation, the very slow diffusion of oxygen in water makes that soil Eh drop rapidly [18].

Surprisingly, the correlation between soil C content and Eh was positive, when soil C is often regarded as the electron reservoir [30], and thus a higher C content is expected to have a lower soil Eh. This suggests that most of the C was under a very stable form, not accessible to microbes and, as a consequence, a raise in total C content does not correspond to the expected raise in biological activity, which would lower the Eh [31]. Again, the hypothesis that most C in these soils is under a very stable form is supported by the frequent fires in the study area, known to produce very stable black C [32,33]. Thus, the limitation of the available nutrients for microbes probably limits biological activity, which could be expected when soil moisture increases. As a consequence, the increase in soil moisture (Figure 10e) does not lead to an increase in soil biological activity sufficient to compensate for the chemical increase in Eh due to a higher water content. Better than total organic matter, it is the bioavailable organic matter, the part of the organic matter accessible to microorganisms, which actually constitutes the main reservoir of electrons [34].

This general framework can explain the variations in Eh, depending on the soil type/location on the toposequence:

- (i) In the upland ecology (Upland 1 and Upland 2), the average redox potential was high (570 mV and 572 mV for Upland 1 and Upland 2, respectively). Eh always remained above 500 mV and the amplitude of variation remained very low for all the soil depths, particularly at a depth of 3 cm. The low variation in soil Eh can be related to the rather high organic C content of these soils, and probably the high content of stable, black C related to frequent fires. Biochar is known to have a high poising capacity for Eh [32,33]. Again, this supports the assumption that these soils have a low biological activity, and that most of their C is stable C. This C and the soil texture retain water, thus increasing soil moisture, while biological activity remains low. The Eh functioning of these soils is mainly related to soil chemistry, with high Eh related to soil moisture.
- (ii) At the fringe, the redox potential always remained higher than 400 mV for all soil depths. This indicates that the soil of the fringe has remained well drained [18]. The lower mean redox potential (556 mV for Fringe 1 and 558 mV for Fringe 2) as compared to upland conditions might be explained by the lower soil moisture, related to lower total soil C. The higher fluctuations in redox potential (as compared to upland soils) may be explained by the lower soil C content, as C, especially black C, is known to poise the redox potential.
- (iii) The glacis had the lowest mean Eh (544 mV) and the highest amplitude in fluctuations, varying from 250 to 714 mV for depths 3, 8, 20 and 35 cm all together. These large amplitudes on glacis can be explained by soil heterogeneity due, on the one hand, to its transitional position between the dry soils, upstream, and on the fringe and the wet soils, downstream. However, first of all, these large amplitudes in Eh on the glacis can be explained by local resurgences of water at this level in relation to a clayey layer. These resurgences of water led to temporary waterlogging, and therefore a strong decrease in soil Eh during the periods of water saturation: in water-saturated soils, oxygen diffusion is very slow, and dissolved oxygen is quickly consumed by soil organisms and plant roots; the soil would then have become anaerobic, as reported by [35], and these reduced conditions would cause a rapid fall in the redox potential. Similar observations have

been made by [18,35,36]. Drainage that occurred a few days after the rain resulted in an increase in soil oxidation, as was also observed by [37]. Declines in the level of soil redox potential were particularly pronounced in mid-October, passing below 400 mV before rising above 500 mV. The reduction of the redox potential was very pronounced at the lower depths, with the lowest soil redox potential recorded at a depth of 20 cm (285 mV).

(iv) Surprisingly, the valley bottom recorded the highest mean redox potential, ranging from 466 to 639 mV, with the highest average redox potential (574 mV) of all sites. This may be explained by the fact that although it is located lower on the toposequence than the glacis, there was no observed flooding of the valley bottoms, even when the glacis was saturated with water from local resurgences. As a consequence, the high soil moisture in a soil rich in organic matter (especially black C), without water saturation, may explain the highest mean redox potential measured in the valley bottoms. The stronger ECb in the wettest parts might also contribute to raise the redox potential, which would also contribute to the higher Eh in the valley bottoms as compared to upland conditions.

Furthermore, the pH generally showed a negative correlation with redox potential, although the observed correlations were low. This can be partly attributed to the variability between sites because better correlations were observed within sites and in particular in the valley bottom.

Variations in pe+pH were low as compared to variations in Eh or pH alone; this parameter being mainly impacted by soil saturation in the glacis and to a lower extent in the valley bottoms.

Finally, it can be hypothesized that the differences in natural vegetation observed before land reclamation were at least partially related to differences in soil electrochemical conditions, as already observed by [38].

5. Conclusions

This study evaluated the spatial and temporal variation in soil redox potential at different positions along the upland–lowland continuum during two growing seasons. Soil redox potential was above 500 mV at the upland positions, between 400 and 700 mV at the fringe positions and between 100 and 750 mV at the lowland positions. Variation in soil redox potential was driven by soil moisture, bulk electrical conductivity and soil organic carbon. The results presented here have several implications for further studies:

Firstly, regarding the protocols for measurement of soil electrochemical parameters: in order to take into account the soil moisture content at the moment of the measurement, it should be systematically measured together with soil Eh, pH and EC.

Secondly, regarding soil carbon content: there is an urgent need to consider not only the total carbon content, but also the carbon easily accessible to microbes as it strongly impacts soil biological activity, which is a major factor regulating soil Eh. Thus, the form of soil carbon should be assessed in future studies dealing with soil electrochemical properties.

Thirdly, regarding the best parameter to assess the soil oxidation–reduction level: The use of pe+pH, which showed lower variability in space and time than Eh or pH alone, could be a good indicator of soil electrochemical conditions.

Finally, although it shows that soil Eh-pH-EC can be accurately be measured with adapted protocols, this study clearly shows the very high spatial and temporal variability of soil Eh, but also pH and ECb. The sensitivity of these parameters to other parameters, especially humidity, raises strong limitations for their use to assess soil health. Even if conducted in fixed conditions of humidity, temperature, etc., a single measurement will not capture their variations nor reflect the actual conditions during the cropping season. A major challenge is to relate these important parameters to the production processes and assess their impact on plant production. To achieve this, Eh-pH-EC should be measured on a very regular basis, with frequent measurements all over the cultivation period, and especially following rainfall events. Doing so would allow to study the elaboration of yield, relating all the yield components to the actual soil Eh-pH-EC conditions during the period of their elaboration. It would

also provide information on the capacity of a soil to buffer soil Eh, pH and EC and thus limit their variation, which appears to be an important parameter related to soil health.

Supplementary Materials: The following are available online at http://www.mdpi.com/2073-4395/10/11/1787/s1, Figure S1. Spatial variability of: soil pH (S1.1), soil electrical conductivity (S1.2), soil carbon (S1.3), soil sand and clay (S1.4), Figure S2: Soil temperature at different soil depths during 2016 and 2017 growing seasons. Error bars indicate standard error of means, Figure S3. Relationship between organic carbon and humidity according to: upland ecologies (S3.1), Fringe ecologies (S3.2) and Lowland ecologies (S3.3), Figure S4. Relationship between pH and Eh according to: upland ecologies (S4.1), fringe ecologies (S4.2), lowland ecologies (S4.3).

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References

- 1. Vanlauwe, B.; AbdelGadir, A.H.; Adewopo, J.; Adjei-Nsiah, S.; Ampadu-Boakye, T.; Asare, R.; Baijukya, F.; Baars, E.; Bekunda, M.; Coyne, D.; et al. Looking back and moving forward: 50 years of soil and soil fertility management research in sub-Saharan Africa. *Int. J. Agric. Sustain.* **2017**, *15*, 613–631. [CrossRef] [PubMed]
- 2. Depieu, E.M.; Doumbia, S.; Keli, J.Z.; Michel, Z. Typologie des exploitations en riziculture pluviale de la région de Saïoua, en zone forestière de la Côte d' Ivoire. *J. Appl. Biosci.* **2010**, *35*, 2260–2278.
- 3. Haefele, S.M.; Saito, K.; N'Diaye, K.M.; Mussgnug, F.; Nelson, A.; Wopereis, M.C.S. Increasing rice productivity through improved nutrient use in Africa. In *Realizing Africa's Rice Promise*; Wopereis, M.C.S., Johnson, D.E., Ahmadi, N., Tollens, E., Jalloh, A., Eds.; CAB International: Wallingford, UK, 2013; pp. 250–264.
- 4. Vlaar, J.C.J. Les Techniques de Conservations des Eaux et des Sols Dans les Pays du Sahel. 1992. Available online: https://www.ircwash.org/sites/default/files/276-92TE-11865.pdf (accessed on 26 September 2019).
- 5. Husson, O. Redox potential (Eh) and pH as drivers of soil/plant/microorganism systems: A transdisciplinary overview pointing to integrative opportunities for agronomy. *Plant Soil* **2013**, *362*, 389–417. [CrossRef]
- 6. Arshad, M.A.; Martin, S. Identifying critical limits for soil quality indicators in agro-ecosystems. *Agric. Ecosyst. Environ.* **2002**, *88*, 153–160. [CrossRef]
- 7. Husson, O.; Audebert, A.; Benada, J.; Soglonou, B.; Tano, F.; Dieng, I.; Bousset, L.; Sarthou, J.-P.; Joseph, S.; Menozzi, P.; et al. Leaf EH and pH: A novel indicator of plant stress. Spatial, Temporal and Genotypic Variability in Rice (*Oryza sativa* L.). *Agronomy* 2018, *8*, 209. [CrossRef]
- 8. Morel, L. Les Sols Cultivés, 2nd ed.; Ed. Tec & Doc-Lavoisier: Paris, France, 1996.
- 9. Petruzzelli, G. Recycling wastes in agriculture: Heavy metal bioavailability. *Agric. Ecosyst. Environ.* **1989**, 27, 493–503. [CrossRef]
- 10. Pigozzo, A.T.J.; Lenzi, E.; De Luca, J.; Scapim, C.A.; Da Costa, A.C.S. Transition metal rates in latosol twice treated with sewage sludge. *Braz. Arch. Biol. Technol.* **2006**, *49*, 515–526. [CrossRef]
- 11. Paillet, Y.; Cassagne, N.; Brun, J.J. Monitoring forest soil properties with electrical resistivity. *Biol. Fertil. Soils* **2010**, *46*, 451–460. [CrossRef]
- 12. Samouëlian, A.; Cousin, I.; Tabbagh, A.; Bruand, A.; Richard, G.; Samoue, A. Electrical resistivity survey in soil science: A review. *Soil Tillage Res.* **2005**, *83*, 173–193. [CrossRef]
- 13. Greenland, D.J.; Hayes, M.H. *The Chemistry of Soil Processes*; Greenland, D.J., Hayes, M.H.B., Eds.; Wiley: Chichester, PA, USA, 1981; Volume 7, pp. 295–296.
- 14. Snakin, V.V.; Prisyazhnaya, A.A.; Kovacs-Lang, E. *Soil Liquid Phase Composition*, 1st ed.; Elsvier: Amsterdam, The Netherlands, 2001; ISBN 0-444-50675-6.

- Husson, O.; Husson, B.; Brunet, A.; Babre, D.; Alary, K.; Sarthou, J.P.; Charpentier, H.; Durand, M.; Benada, J.; Henry, M. Practical improvements in soil redox potential (Eh) measurement for characterisation of soil properties. Application for comparison of conventional and conservation agriculture cropping systems. *Anal. Chim. Acta* 2016, 906, 98–109. [CrossRef]
- 16. Chesworth, W.; Cortizas, A.M.; Garcı, E. The redox—pH approach to the geochemistry of the Earth's land surface, with application to peatlands. *Dev. Earth Surf. Process.* **2006**, *2025*, 175–195.
- 17. Schwertmann, U.; Murad, E. Effect of pH on the formation of goethite and hematite from ferrihydrite. *Clays Clay Miner.* **1983**, *31*, 277–284. [CrossRef]
- 18. Ponnamperuma, F.N. The chemistry of submerged soils. In *Advances in Agronomy*; Academic Press: Cambridge, MA, USA, 1972; Volume 24, pp. 29–96.
- Hall, S.J.; Mcdowell, W.H.; Silver, W.L. When Wet Gets Wetter: Decoupling of Moisture, Redox Biogeochemistry, and Greenhouse Gas Fluxes in a Humid Tropical Forest Soil. *Ecosystems* 2013, 16, 576–589. [CrossRef]
- 20. Stumm, W.; Morgan, J.J. An introduction emphasizing chemical equilibria in natural waters. In *Aquatic Chemistry*; Wiley-Interscience: New York, NY, USA, 1981; pp. 282–285. ISBN 978-0-471-51185-4.
- 21. Szafranek-nakonieczna, A.; Stepniewska, Z. The influence of the aeration status (ODR, Eh) of peat soils on their ability to produce methane. *Wetl. Ecol. Manag.* **2015**, *23*, 665–676. [CrossRef]
- Diatta, S. Les Sols Gris de Bas Versant sur Granito-Gneiss en Région Centrale de la Côte d'Ivoire: Organisation Toposéquentielle et Spatiale, Fonctionnement Hydrologique: Conséquences Pour la Riziculture. Ph.D. Thesis, Université Henri Poincaré-Nancy 1, Nancy, France, 1996.
- 23. NF X31-107. *Qualité du Sol-Détermination de la Distribution Granulométrique des Particules du Sol-Méthode à la Pipette;* AFNOR: Paris, France, 2003.
- 24. Saito, K.; Asai, H.; Zhao, D.; Laborte, A.G.; Grenier, C. Progress in varietal improvement for increasing upland rice productivity in the tropics. *Plant Prod. Sci.* **2018**, *21*, 145–158. [CrossRef]
- 25. Pidello, A. Environmental redox potential and redox capacity concepts using a simple polarographic experiment. *J. Chem. Educ.* **2003**, *80*, 68–70. [CrossRef]
- 26. Lindsay, W.L.; Sadiq, M. Use of pe+pH to predict and interpret metal solubility relationships in soils. *Sci. Total Environ.* **1983**, *28*, 169–178. [CrossRef]
- 27. Patrick, W.H.J.; Mahapatra, I.C. Transformation and availability to rice plant of nitrogen and phosphorus in waterlogged soils. *Adv. Agron.* **1968**, *20*, 323–359.
- 28. Bohrerova, Z.; Stralkova, R.; Podesvova, J.; Bohrer, G.; Pokorny, E. The relationship between redox potential and nitrification under different sequences of crop rotations. *Soil Tillage Res.* **2004**, *77*, 25–33. [CrossRef]
- 29. Kralova, M. Redox Potential as an Indicator of Electron Availability for Microbial. *Zentralbl. Mikrobiol.* **1992**, 147, 388–399. [CrossRef]
- 30. Chesworth, W. Redox, soil and carbon sequestration. Edafologia 2004, 11, 37–43.
- 31. Lovley, D.R.; Fraga, J.L.; Blunt-Harris, E.L.; Hayes, L.A.; Phillips, E.J.P.; Coates, J.D. Humic substances as a mediator for microbially catalyzed metal reduction. *Acta Hydrochim. Hydrobiol.* **1998**, *26*, 152–157. [CrossRef]
- 32. Joseph, S.; Husson, O.; Graber, E.R.; Van Zwieten, L.; Taherymoosavi, S.; Thomas, T.; Nielsen, S.; Ye, J.; Pan, G.; Chia, C.; et al. The electrochemical properties of biochars and how they affect soil redox properties and processes. *Agronomy* **2015**, *5*, 322–340. [CrossRef]
- Klüpfel, L.; Keiluweit, M.; Kleber, M.; Sander, M.; Kluepfel, L.; Keiluweit, M.; Kleber, M.; Sander, M. Redox properties of plant biomass-derived black carbon (biochar). *Environ. Sci. Technol.* 2014, 48, 5601–5611. [CrossRef] [PubMed]
- 34. Chadwick, O.A.; Chorover, J. The chemistry of pedogenic thresholds. *Geoderma* 2001, 100, 321–353. [CrossRef]
- 35. Kyuma, K.P. Paddy Soil Science; Kyoto University Press: Kyoto, Japan, 2004; pp. 273–280. ISBN 1920901000.
- 36. Vizier, J.F. Etude de l'état d'oxydoréduction du sol et de ses conséquences sur la dynamique du fer dans les sols hydromorphes. *Cah. ORSTOM* **1971**, *9*, 373–397.

- 37. Kongchum, M. Effect of Plant Residue and Water Management Practices on Soil Redox Chemistry, Methane Emission, and Rice Productivity. Ph.D. Thesis, Louisiana State University, Baton Rouge, LA, USA, 2005.
- 38. Pennington, M.R.; Walters, M.B. The response of planted trees to vegetation zonation and soil redox potential in created wetlands. *Forest Ecol. Manag.* **2006**, *233*, 1–10. [CrossRef]

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